TOTAL SYNTHESIS OF NEOMYCIN C

Sir:

Neomycin¹⁾, a useful antibiotic, was independently discovered by H. UMEZAWA *et al.*²⁾ in 1948 and S. A. WAKSMAN *et al.*¹⁾ in 1949. The neomycin complex consists of neomycins A, B, and C and neomycin C was isolated by DUT-CHER *et al.*³⁾ Elucidation of the complete structures of neomycins was achieved by RINEHART and his associates^{4,5)} in 1963. We wish here to report the total synthesis of neomycin C (1). This represents the first synthesis of an antibiotic of pseudo-tetrasaccharides.

Neomycin C (1) is composed of two molecules of neosamine C (2,6-diamino-2,6-dideoxy-Dglucopyranose)⁶⁾, 2-deoxystreptamine⁷⁾, and Dribose. Neamine (neomycin A), which is a pseudodisaccharide component of neomycins B and C, was synthesized by S. UMEZAWA *et al.*⁸⁾ and ribosylation of neamine was subsequently reported by ITO *et al.*⁹⁾ to afford ribostamycin which corresponds to a pseudo-trisaccharide portion common to both neomycins B and C. Hence, our approach to the synthesis of neomycin C involved pertinent protection of neosamine C and ribostamycin and regioselective α -glycosidation between them.

The α -glycosidation reaction used in the present study is based on our prior findings^{10,11}) that the modified KOENIGS-KNORR condensation of 3,4,6-tri-O-acetyl-2-deoxy-2-(*p*-methoxybenzyl-ideneamino)- α -D (or L)-glucopyranosyl bromide with suitably substituted sugars gave high yields (~85%) of α -glycosides. Our initial attempt to brominate 1,3,4-tri-O-acetyl-6-azido-2, 6-dideoxy-2-(*p*-methoxybenzylideneamino)- β -D-glucopyranose (**2**, mp 115~116°C, [α]_D²⁰ + 117.6°

(c 1.43, chloroform)) with hydrogen bromide in methylene chloride (2 minutes at -20° C) was not successful and resulted in the formation of 1, 3,4-tri-O-acetyl-6-bromo-2,6dideoxy-2-(*p*-methoxybenzylideneamino)- β -D-glucopyranose (3,mp151~152°C, $[\alpha]_{D}^{20}+96.8^{\circ}$ (*c* 0.67, chloroform)), the azido group at C-6 being replaced by bromine.¹²⁾ In a second approach to prepare the glycosyl halide, 1,3,4-tri-O- acetyl-2-deoxy-2-(*p*-methoxybenzylideneamino)-6-O-tosyl- β -D-glucopyranose¹⁸) was successfully converted, in 85% yield, by reaction of hydrogen bromide in methylene chloride (3 hours at 0°C), to syrupy 3,4-di-O-acetyl-2-deoxy-2-(*p*-methoxybenzylideneamino)-6-O-tosyl- α -Dglucopyranosyl bromide (4). Its structure was confirmed by ¹H-NMR data (in CDCl₃): δ 1.83, 1.95, and 2.43 (each 3H, s, OAc), 3.43 (1H, q, J_{1,2}=4.5 Hz and J_{2,3}=9 Hz, C₂-H), 3.85 (3H, s, OCH₃), 6.15 (1H, d, J_{1,2}=4.5 Hz, C₁-H), 6.90 and 7.68 (4H, AA'BB', J₀=9 Hz, C₆<u>H</u>₄OCH₃), 7.35 and 7.82 (4H, AA'BB', J₀= 9 Hz, SO₂C₆H₄CH₃).

On the other hand, ribostamycin was protected in the following manner. 1,3,2',6'-Tetra-Nbenzyloxycarbonylribostamycin¹⁴⁾ was transformed to the 2'',3''-O-cyclohexylidene ketal (5), mp $119 \sim 120^{\circ}$ C, $[\alpha]_{D}^{20} + 6^{\circ}$ (*c* 1.0, chloroform), in 80% yield, by selective cyclohexylidenation







(10) $R^{1} = NHCO_2 CH_2 Ph$, $R^{2} = N_3$

with an excess of 1,1-dimethoxycyclohexane (~2 mol) in N,N-dimethylformamide (20 hours at 70°C) in the presence of *p*-toluenesulfonic acid. The ketal was treated with acetic anhydride and pyridine (overnight at room temperature) to give the tetraacetyl derivative (**6**), mp 101~ 102°C, $[\alpha]_D^{21}$ +16.2° (*c* 1.54, chloroform), in 98% yield. Partial hydrolysis of **6** with 50% acetic acid (12 hours at 80°C) provided, after column chromatography on silica gel, 6,3',4',5''-tetra-O-acetyl-1,3,2',6'-tetra-N-benzyloxycarbonylribostamycin (**7**), mp 102~103°C, $[\alpha]_D^{21}$ +18° (*c* 1.39, chloroform), in 35% yield.

Reaction of the protected glycosyl halide (4) with 7 in anhydrous chloroform in the presence of silver carbonate, silver perchlorate and Drierite (20 hours at room temperature) followed by column chromatography on silica gel, afforded a mixture (a syrup) of the condensation products containing 8 as the main product. Hydrolysis of the SCHIFF base moiety of the condensation products at pH 2 in acetone-hydrochloric acid (several minutes at room temperature) gave a syrup. The free amino group at C-2" was benzyloxycarbonylated by treatment with benzyloxycarbonyl chloride in aqueous acetone in the presence of sodium carbonate (20 hours at room temperature). Silica gel chromatography of the reaction mixture afforded 6,3',4',5"-tetra-O-acetyl-1,3,2',6'-tetra-N-benzyloxycarbonyl- 3"-O-(3,4-di-O-acetyl-2-benzyloxycarbonylamino-2deoxy-6-O-tosyl- α -D-glucopyranosyl)ribostamycin (9), a colorless solid, $[\alpha]_{D}^{20} + 42.2^{\circ}$ (c 0.97, chloroform), in 34% overall yield from 7.

Treatment of 9 with sodium azide in N,Ndimethylformamide (15 minutes at 130°C) gave the 6^{'''}-azide (10) as a solid, $[\alpha]_{D}^{20}$ +64.4° (c 0.71,

Table 1.	Antibacterial	spectra	of	synthetic	and
natural	neomycin C				

Test succeives*	MIC (mcg/ml)		
Test organisms*	Synthetic	Natural	
Staphylococcus aureus FDA 209P	3.12	3.12	
Bacillus subtilis NRRL B. 558	0.2	0.39	
Escherichia coli K-12	6.25	6.25	
Escherichia coli K-12 ML 1629	> 100	>100	
Escherichia coli K-12 LA 290 R 55	>100	>100	
Escherichia coli K-12 LA 290 R 56	3.12	6.25	
Escherichia coli K-12 W 677	1.56	1.56	
Escherichia coli K-12 JR 66/W 677	> 100	>100	
Klebsiella pneumoniae PCI 602	3.12	3.12	
Klebsiella pneumoniae type 22 #3038	>100	>100	
Mycobacterium smegmatis ATCC 607**	1.56	1 .5 5	

* Agar dilution streak method (nutrient agar, 37°C, 17 hours).

** 48 hours.

acetone), in 85% yield. De-benzyloxycarbonylation of 10 by catalytic hydrogenolysis with palladium black and hydrogen (4 kg/cm², 14 hours at room temperature) in aqueous dioxane in the presence of acetic acid, followed by hydrolysis with 1 N barium hydroxide (5 hours at 95°C) and purification by use of Sephadex C-25 (NH4⁺) completed the synthesis, giving neomycin C (1), [hydrochloride: $[\alpha]_{\rm D}^{20} + 78^{\circ}$ (c 0.86, water)]¹⁵, identical with that obtained from natural source as judged by the IR spectrum of the hydrochloride (KBr), ¹H-NMR (D₂O) (anomeric protons: $\delta 5.52(1H, d, J_1''', 2''' = 3.5 Hz$, $C_1^{\prime\prime\prime}$ -H), 5.88 (1H, d, $J_1^{\prime\prime},_2^{\prime\prime}$ =1 Hz, $C_1^{\prime\prime}$ -H), and 5.94 (1H, d, $J_{1',2}'=3$ Hz, $C_{1'}$ -H), and ${}^{13}C$ -The tentative assignment of NMR spectra. the ¹³C-NMR spectra (25.16 MHz, D₂O, $\delta =$ internal dioxane difference (67.4 ppm)): 36.4; 42.4; 42.5; 51.1 (2 carbons); 55.9; 56.3; 62.4; 72.2 (2 carbons); 73.98 (4 carbons); 74.5; 75.5; 78.4, 81.7; 82.8; 85.2; 99.7 (C1' and C1'''); 109.45 (C_1'') . Identity with the natural product was further indicated by the antibacterial spectra shown in Table 1.

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 3). [We found [α]²⁰₂+79° (c 0.86, water).]